

ANTITUMOR AGENTS, 110.^{1,2} BRYOPHYLLIN B, A NOVEL POTENT CYTOTOXIC BUFADIENOLIDE FROM *BRYOPHYLLUM PINNATUM*

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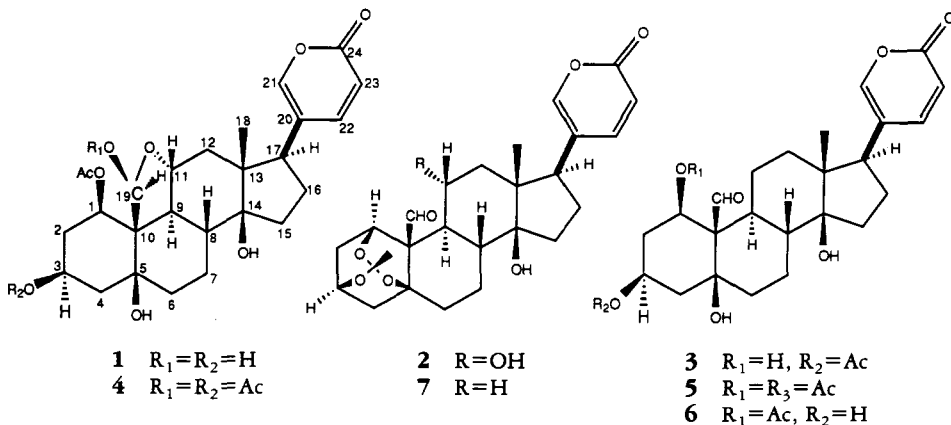
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ABSTRACT.—Bryophyllin B [**1**], a potent cytotoxic bufadienolide, has been isolated from *Bryophyllum pinnatum* and its structure confirmed by the use of 2D-nmr techniques and difference nOe method. Transformation of bryotoxin C [**2**] to **1** with acid is also discussed.

We reported previously on the isolation of bryotoxin C (bryophyllin A) [**2**] (1), a bufadienolide 1,3,5-orthoacetate with potent cytotoxicity, and bersaldegenin-3-acetate [**3**] (3,4) from the CHCl₃ extract of *Bryophyllum pinnatum* (Crassulaceae). Bryotoxin C was first isolated by Capon *et al.* (2) from *Bryophyllum tubiflorum* without reporting any biological activity. Further investigation of a cytotoxic H₂O extract of this same plant has led to the isolation of bryophyllin B [**1**], which showed potent cytotoxicity with ED₅₀ < 80 ng/ml against the in vitro growth of KB tissue culture cells (Table 1). We report herein the isolation and structural elucidation of bryophyllin B [**1**]. The transformation of **2** to **1** with acid is also discussed.

RESULTS AND DISCUSSION

The H₂O extract of the whole plant of *B. pinnatum* was concentrated and partitioned between H₂O and CHCl₃. Guided by the assay in KB cells as shown in Scheme 1, the active principles were concentrated in the CHCl₃ (Fractions A and B) and the H₂O extracts. Bryotoxin C (bryophyllin A) [**2**] and bersaldegenin-3-acetate [**3**] were isolated from the CHCl₃ extract. The H₂O-soluble part was extracted with *n*-BuOH, which in turn was chromatographed on Sephadex LH-20 Si gel and reversed-phase



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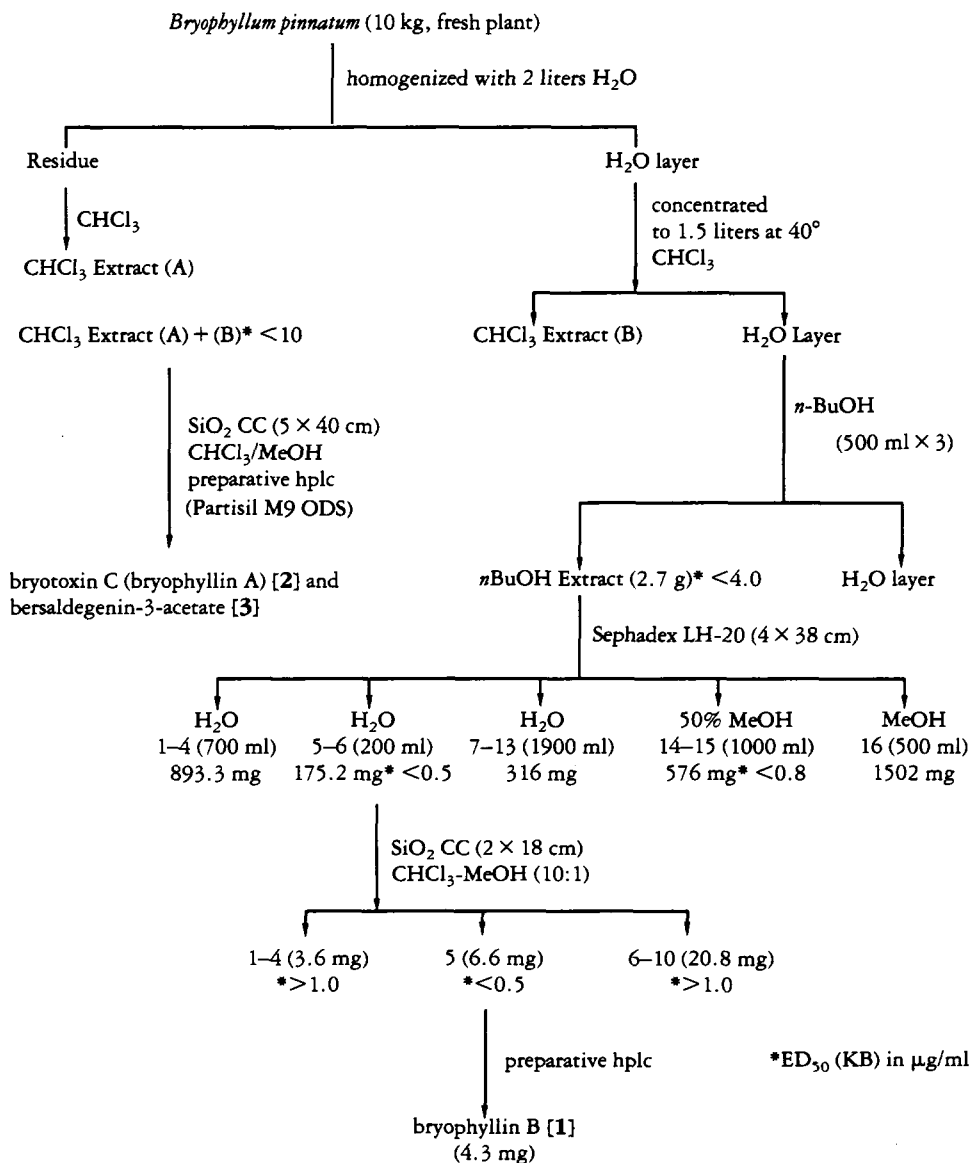
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TABLE 1. Cytotoxicity of Compounds **1**, **2**, and **3** Against Various Tumor Cells.

Compound	KB (ng/ml)	A-549 (ng/ml)	HCT-8 (ng/ml)	P-388 (μ g/ml)	L-1210 (μ g/ml)
Bryophyllin B [1]	<80	—	—	—	—
Bryotoxin C [2]	14	10	30	>4	>4
Bersaldegennin-3-acetate [3]	<40	40	10	>4	>4

hplc, successively, to afford the active principle, bryophyllin B [**1**], in 0.000043% yield.

Bryophyllin B [**1**] was obtained as a colorless amorphous powder and analyzed for $C_{26}H_{34}O_9$. Its uv and ir spectra indicated the presence of a dienone system (λ max 298

SCHEME 1. Extraction and Isolation of Bryophyllin B [**1**].

nm) as well as hydroxyl (3400 cm^{-1}) and carbonyl (1695 cm^{-1}) groups as seen in **2**. The fabms of **1** showed the appearance of an $[M]^+$ peak at m/z 490 and fragment ion peaks at m/z 472 $[M - H_2O]^+$ and 432 $[M - HOAc]^+$.

Detailed analysis of the ^1H - and ^{13}C -nmr spectra (Table 2) of **1** in ^1H - ^1H COSY (Correlation Spectroscopy) (Figure 1) and ^{13}C - ^1H COSY (Figure 2) suggested the presence of a methyl (δ_{H} 0.84 and δ_{C} 20.57), a secondary acetoxy (δ_{H} 2.07 and 4.69; δ_{C} 171.00, 21.20, and 73.38), a γ -substituted $\alpha,\beta,\gamma,\delta$ -unsaturated- γ -lactone (α -pyrone) (δ_{H} 7.45, 7.87, and 6.92; δ_{C} 149.32, 147.32, 115.41, 122.66, and 162.07), a secondary hydroxyl (δ_{H} 3.79 and δ_{C} 65.01), two tertiary hydroxyl (δ_{C} 76.08 and 85.60), and lactol (δ_{H} 5.75, 3.99, and 1.30; δ_{C} 104.66, 79.99, and 49.53) groups. Treatment of **1** with Ac_2O in pyridine gave diacetate **4**, which was transformed to monoacetate **5** after standing in CHCl_3 for two days. These foregoing data and experiment indicate that **1** may be a compound closely related to **2** except that the orthoacetate and acetal groups differ from each other.

Each carbon signal, except the quaternary one, was assigned based on the ^{13}C - ^1H COSY spectral data (Figure 2). In the ^1H - ^1H COSY spectrum of **1**, six mutual relations from H-1, H-3 (twice), H-8, H-11, and H-17 to 2- CH_2 , 4- CH_2 , 7- CH_2 , 12- CH_2 , and 16- CH_2 methylene protons due to an ABX system were elucidated as shown in Figure 3. The configuration of each X-type proton of this ABX system is in an axial orientation, which has a large coupling constant (Table 1). Furthermore, the relationships be-

TABLE 2. ^1H and ^{13}C Chemical Shifts of Compound **1**.^a

Position	^1H	^{13}C
1	4.69 (1H, dd, $J = 4.03, 12.45$ Hz)	73.38 (d)
2	2.11 (1H, ddd, $J = 4.03, 4.64, 12.45$ Hz) 1.54 (1H, q, $J = 12.45$ Hz)	39.51 (t)
3	3.79 (1H, m)	65.01 (d)
4	1.96–1.98 and 1.88 (each 1H, obscured signal)	47.79 (t)
5		76.08 (s)
6	1.98 and 1.58 (each 1H, obscured signal)	32.11 (t)
7	1.11 (1H, dddd, $J = 3.78, 9.52, 12.94, 13.31$ Hz) 1.96–1.98 (1H, obscured signal)	20.78 (t)
8	2.42 (1H, ddd, $J = 2.93, 11.35, 13.31$ Hz)	39.97 (d)
9	1.30 (1H, t, $J = 11.35$ Hz)	49.53 (d)
10		54.37 (s)
11	3.99 (1H, ddd, $J = 3.30, 11.35, 11.85$ Hz)	79.99 (d)
12	1.46 (1H, t, $J = 11.85$ Hz) 2.03 (1H, dd, $J = 3.30, 11.85$ Hz)	48.22 (t)
13		52.63 (s)
14		85.60 (s)
15	1.96–1.98 and 1.56 (each 1H, obscured signal)	34.92 (t)
16	2.21 and 1.83 (each 1H, m)	31.10 (t)
17	2.71 (1H, dd, $J = 7.57, 8.79$ Hz)	51.23 (d)
18	0.84 (3H, s)	20.57 (q)
19	5.75 (1H, s)	104.66 (d)
20		122.66 (d)
21	7.45 (1H, m)	149.32 (d)
22	7.87 (1H, dd, $J = 2.57, 9.77$ Hz)	147.32 (d)
23	6.29 (1H, dd, $J = 0.85, 9.77$ Hz)	115.41 (d)
24		162.07 (s)
-COMe	2.07 (3H, s)	21.20 (q)
-COMe		171.00 (s)

^aThe measurements were made on a JEOL GX-400 spectrometer in CD_3OD with TMS as an internal reference, and are expressed in terms of ppm.

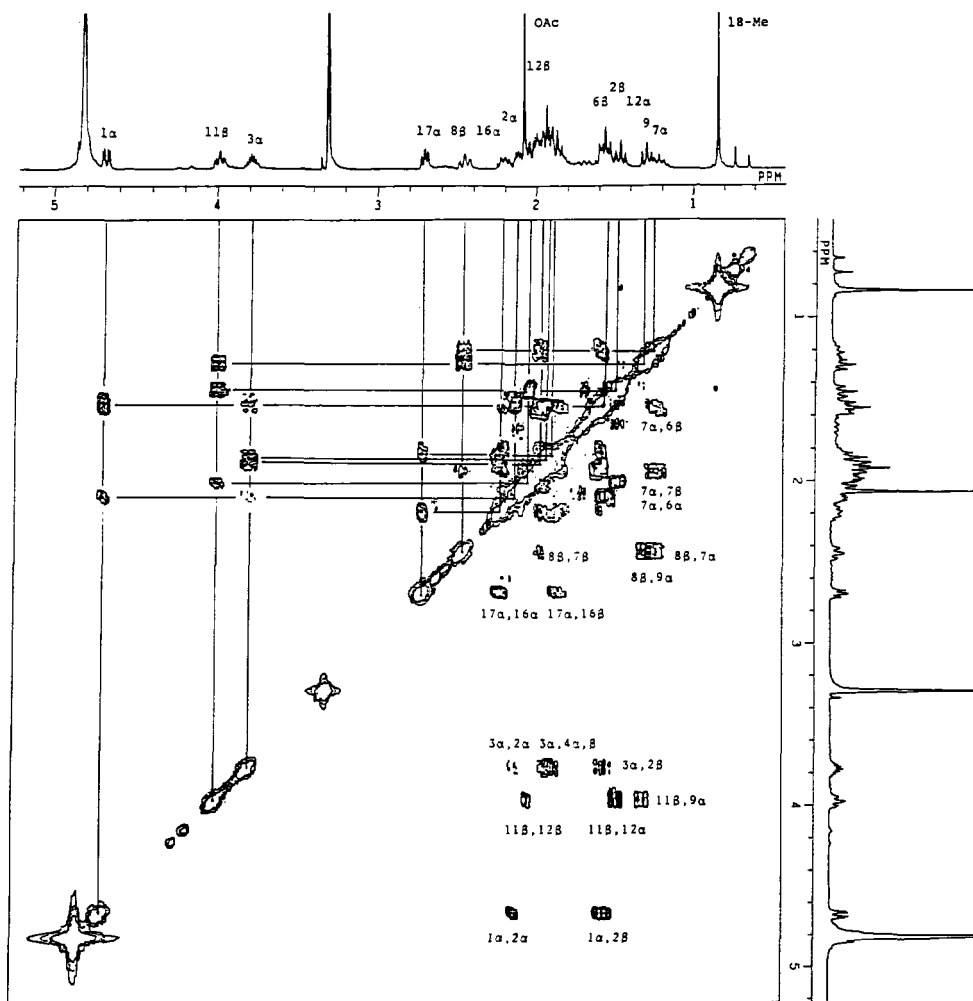


FIGURE 1. ^1H - ^1H COSY Spectrum of Bryophyllin B [1].

tween H-8 and H-9 and between H-9 and H-11 were also trans diaxial ($J_{8,9}$ and $J_{9,11} = 11.35$ Hz) to each other.

The ^{13}C - ^1H long range COSY (5) of **1** was measured in order to confirm the partial structure discussed below. As shown in Figure 3, the proton signal at δ 0.84 (Me-18) is correlated with the carbons at δ 48.22 (C-12), 52.63 (C-13), 85.60 (C-14), and 51.23 (C-17), and the signal at δ 1.30 (H-9) is correlated with the carbons at δ 76.08 (C-5) and 54.37 (C-10). Also, the proton signals at δ 7.45 (H-21), 7.87 (H-22), and 6.29 (H-23) are correlated with the carbons at δ 122.66 (C-20), 147.32 (C-22) and 162.07 (C-24), and δ 162.07 (C-24) and δ 122.66 (C-20), respectively. Furthermore, the proton signal at δ 4.69 (H-1) is correlated with the carbon at δ 104.66 (C-19), suggesting that the position of the acetoxy group is at C-1. Some of the other significant long-range correlations observed are shown by arrows in Figure 3.

The relative stereochemistry of **1** and connectivity between C-17 and C-20 were determined on the basis of coupling constants of each proton and the results of difference nOe experiments (Figure 4). Irradiation at the frequency of the Me-18 proton signal (δ 0.84) enhanced the signal intensity of two olefins (H-21 and H-22), one of the methylene (H-12), and two methine protons (H-8 and H-11). This suggests that the

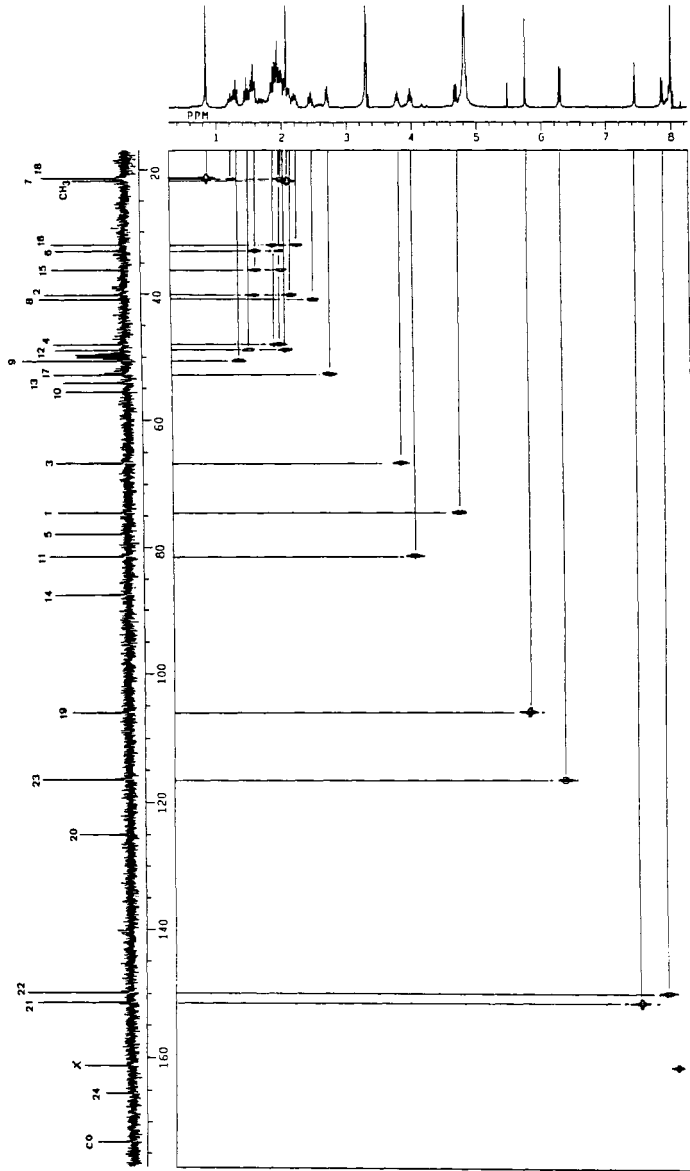


FIGURE 2. ^{13}C - ^1H COSY Spectrum of Bryophyllin B [1].

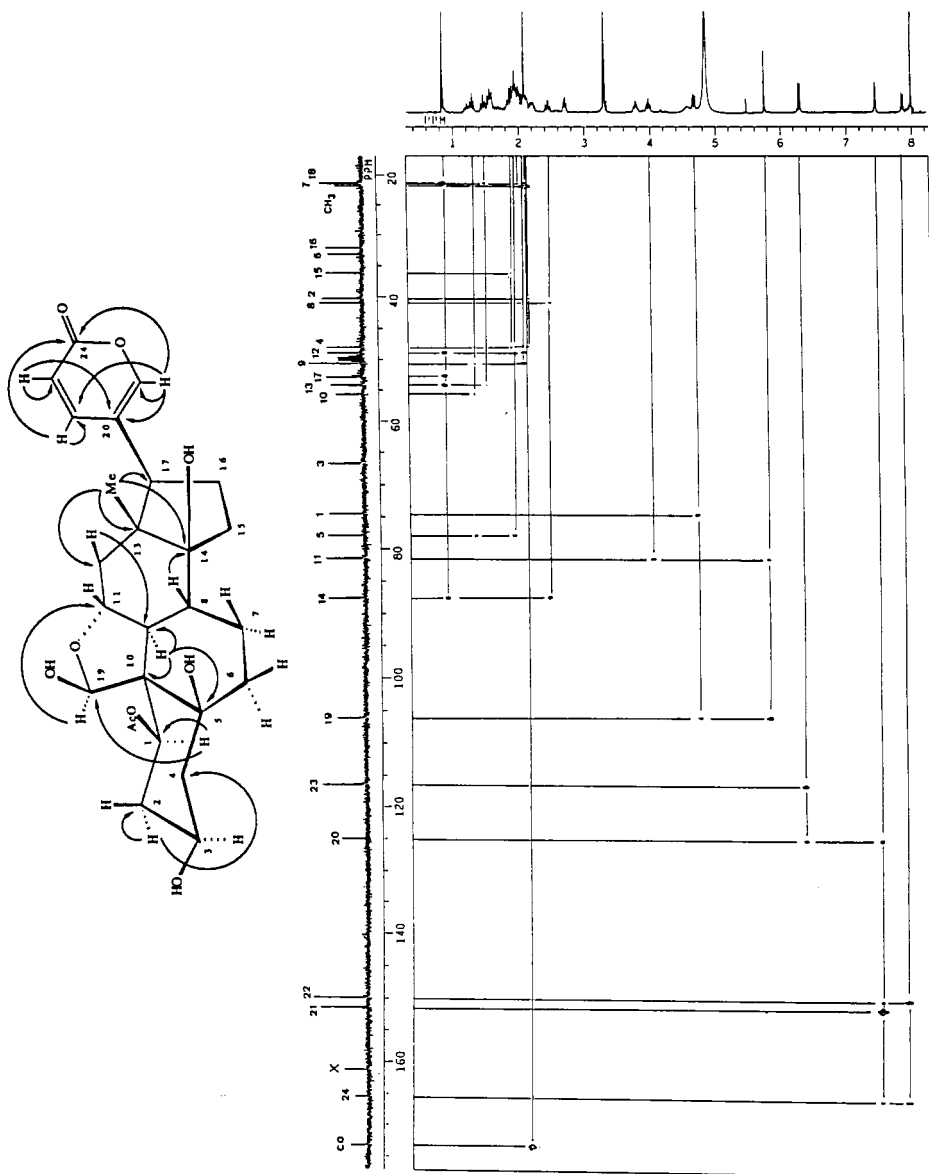


FIGURE 3. Long-range ^{13}C - ^1H COSY of Bryophyllin B [1] ($J = 10.0$ Hz).

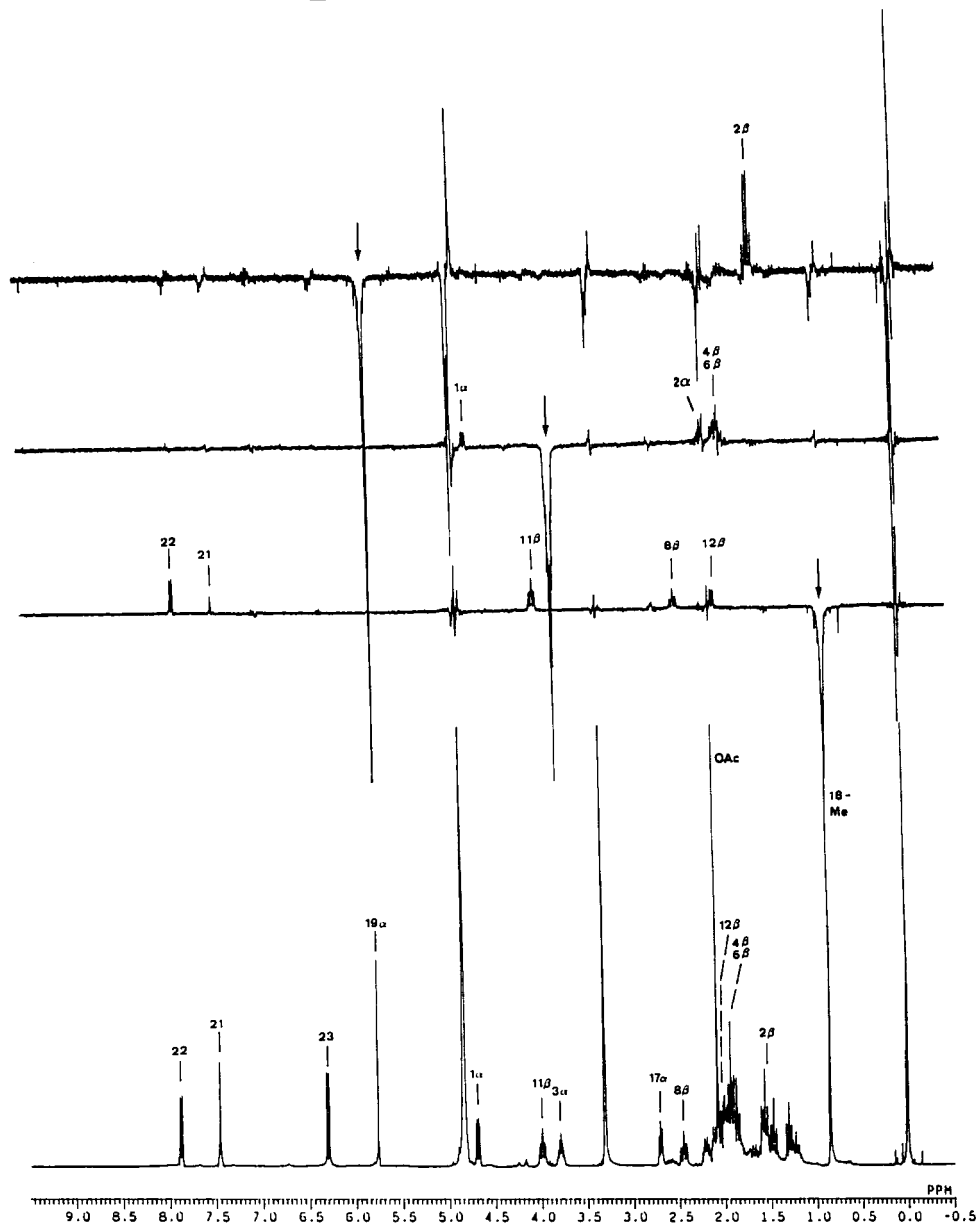
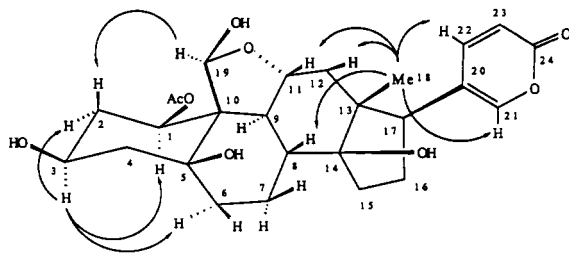


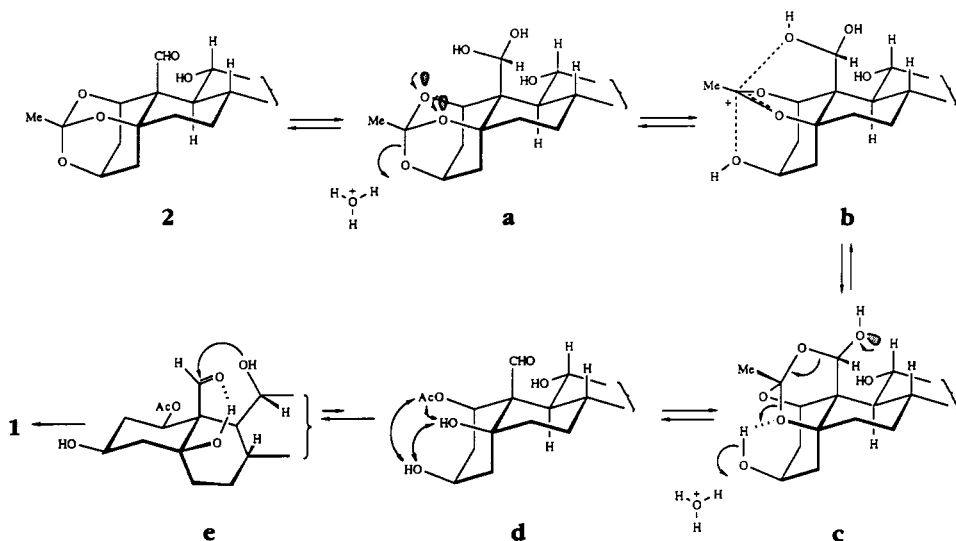
FIGURE 4. Difference nOe of Bryophyllin B [1]

Me-18, 8-, and 11-protons have a 1,3-diaxial relationship to each other and that the Me-18 and α -pyrone ring are in a *cis* orientation. Irradiation at the frequency of H-3 (δ 3.79) and H-19 (δ 5.75) enhanced the signal intensity of H-1 (δ 4.69), H-6 (δ 1.98), and the methylene proton (H-2 β , δ 1.54).

Based on the above evidence and a biogenetic point of view regarding the co-occurrence of bryophyllin B with **2** and **3** from the same plant, the structure of bryophyllin B was determined to be **1**.

Added confirmation of the structure of **1** was achieved by the transformation of **2** to the 1 β -acetate [**1**]. Compound **2** was treated with 10-camphor sulfonic acid in a mixture of CH₂Cl₂ and H₂O for 2 days at room temperature to afford a 2:1 mixture of **1** and **2**. This type of equilibrium was observed by Kupchan *et al.* (6) previously in their treatment of either bersaldegenin 1-acetate [**6**] or bersaldegenin 1,3,5-orthoacetate [**7**] with 80% HOAc at 90° to yield a 1:1 equilibrium mixture of **6** and **7**.

The formation of **1** from **2** under acidic conditions as described above may be controlled by stereoelectronic effects (7) as explained in Scheme 2.



SCHEME 2. Transformation of **2** to **1**.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—All melting points were taken on a Fisher-Johns melting-point apparatus and are uncorrected. Ir spectra were recorded on a Perkin-Elmer 1320 spectrophotometer. ¹H- and ¹³C-nmr spectra were measured on JEOL GX-400 and Varian XL-400 spectrometers with TMS as an internal standard. Mass spectra were taken on a Shimadzu DF 2000 spectrometer by the fab method. Si gel (Kieselgel 60, 230–400 mesh, Merck) was used for cc, and pre-coated Si gel plates (Kieselgel 60 F₂₅₄, 0.25 mm, Merck) were used for analytical tlc. Detection of bufadienolides was made by spraying with 10% H₂SO₄ solution containing 1% Ce(SO₄)₂ followed by heating. Hplc was carried out on a Waters Associates Model ALC/GPC 244 liquid chromatograph with a 450 variable wavelength detector. The column used in this system was Nucleosil 7C₁₈ (Macherey-Nagel) 10 × 250 mm. MeOH-H₂O-HOAc (60:40:0.2) and MeCN-H₂O-HOAc (40:60:0.2) were used as the mobile phase, and the flow rate was 2–4 ml/min.

PLANT MATERIAL.—*B. pinnatum* (8) was collected in the spring of 1987 in Taipei, Taiwan. A voucher specimen of this plant is kept at the Institute of Botany, Academia Sinica, Taiwan.

EXTRACTION AND ISOLATION OF BRYOPHYLLIN B.—As shown in Scheme 1, the whole fresh plant of *B. pinnatum* (10 kg), was homogenized with H₂O (2 liters). This H₂O extract was filtered and concentrated in vacuo to 1.5 liters. After having been shaken with CHCl₃, it was extracted with *n*-BuOH (500

ml \times 3). The *n*-BuOH extract (2.7 g) was subjected to cc on Sephadex LH-20 (4 \times 38 cm) and eluted with a gradient of H₂O, 50% MeOH, and MeOH to give 16 fractions. Fractions 5, 6, 14, and 15 were found to show significant cytotoxicity in KB cells. Fraction 5 (119.6 mg) was further chromatographed on Si gel (2 \times 18 cm) and eluted with CHCl₃-MeOH (10:1, each fraction 30 ml) to afford 10 fractions. Purification of the active fraction 5 (6.6 mg) by reversed-phase hplc furnished bryophyllin B [**1**] (4.3 mg).

BRYOPHYLLIN B [1].—Bryophyllin B was isolated as a colorless amorphous powder: C₂₆H₃₄O₉; mp 178–180°, [α]²⁰_D +20° (c = 0.1, CHCl₃); uv λ max (MeOH) 298 (ϵ 5800) nm; ir (CHCl₃) 3400 (OH), 1695 (C=O), 1120 (C-O) cm⁻¹; fabms *m/z* [M]⁺ 490, [M - H₂O]⁺ 472, [M - HOAc]⁺ 432.

ACETYLATION OF BRYOPHYLLIN B [1].—A solution of **1** (2.0 mg) in a mixture of pyridine (0.2 ml) and Ac₂O (0.1 ml) was allowed to stand at room temperature overnight, then diluted with H₂O and extracted with EtOAc. The EtOAc extract was washed with saturated NaCl solution, dried over Na₂SO₄, and evaporated. The residue was purified by preparative tlc to give a diacetate **4**: ¹H nmr (CDCl₃) δ 7.67 (1H, dd, J = 2.57, 9.71 Hz, H-22), 7.23 (1H, br s, H-21), 6.74 (1H, s, H-19 α), 6.29 (1H, dd, J = 0.73, 9.71 Hz, H-23), 4.89 (1H, m, H-3 α), 4.74 (1H, dd, J = 4.22, 12.46 Hz, H-1 α), 3.96 (1H, ddd, J = 3.30, 11.17, 11.54 Hz, H-11 β), 2.60 (1H, dd, J = 8.14, 8.80 Hz, H-17 α), 2.43 (1H, br t, J = 11.73 Hz, H-8 β), 2.22 (3H, s, 19 β -OAc), 2.12 (3H, s, 3 β -OAc), 2.03 (3H, s, 1 β -OAc), 0.85 (3H, s, Me-18).

When compound **4** was dissolved in CDCl₃ and allowed to stand for two days, it yielded a monoacetate **5**: ¹H nmr (CDCl₃) δ 7.67 (1H, dd, J = 2.57, 9.71 Hz, H-22), 7.23 (1H, br s, H-21), 6.29 (1H, dd, J = 0.73, 9.71 Hz, H-23), 5.84 (1H, s, H-19 α), 4.89 (1H, m, H-3 α), 4.74 (1H, dd, J = 4.22, 12.46 Hz, H-1 α), 4.05 (1H, ddd, J = 3.30, 11.17, 11.54 Hz, H-11 β), 2.60 (1H, dd, J = 8.14 and 8.80 Hz, H-17 α), 2.43 (1H, br t, J = 11.73 Hz, H-8 β), 2.09 (3H, s, 3 β -OAc), 2.03 (3H, s, 1 β -OAc), 0.85 (3H, s, Me-18).

TRANSFORMATION OF BRYOTOXIN C [2] TO BRYOPHYLLIN B [1].—A solution of **2** (1.0 mg) in a mixture of CH₂Cl₂ (1.0 ml) and H₂O (1 drop) was treated with a catalytic amount of (\pm)-10-camphor sulfonic acid for 2 days at room temperature. The reaction mixture was diluted with H₂O (3.0 ml) and extracted with CH₂Cl₂. The CH₂Cl₂ extract was washed with saturated NaCl and evaporated to yield a residue. This residue showed two peaks in a ratio of 1:2, corresponding to **1** and **2**, respectively, on reversed-phase hplc [7C₁₈, 10 \times 250 mm, MeOH-H₂O-HOAc (50:50:0.1), 3 ml/min, 298 nm].

CYTOTOXICITY ASSAY.—The in vitro KB cytotoxicity assay was carried out according to procedures described in Geran *et al.* (9) and Ferguson *et al.* (10). The assay against A-549, HCT-8, P-388, and L-1210 tumor cells was based on a method reported in Lee *et al.* (11).

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LITERATURE CITED

1. T. Yamagishi, X.Z. Yan, R.Y. Wu, D.R. McPhail, A.T. McPhail, and K.H. Lee, *Chem. Pharm. Bull.*, **36**, 1615 (1988).
2. R.J. Capon, J.K. Macleod, and P.B. Oelrichs, *Aust. J. Chem.*, **39**, 1711 (1986).
3. S.M. Kupchan and J. Ognyanov, *Tetrahedron Lett.*, **21**, 1709 (1969).
4. H. Wagner, H. Lotter, and M. Fisher, *Helv. Chim. Acta*, **69**, 359 (1986).
5. C. Francisco, B. Banaigs, and J. Teste, *J. Org. Chem.*, **51**, 1115 (1986).
6. S.M. Kupchan, I. Ognyanov, and J.L. Moniot, *Bioorg. Chem.*, **1**, 13 (1971).
7. P. Deslongchamps, "Stereolectronic Effects in Organic Chemistry," Pergamon Press, New York, 1983, Chapters 3 and 8.
8. H.L. Li, T.S. Liu, T.C. Huang, T. Koyama, and C.E. DeVol, "Flora of Taiwan," (1977), Vol. 3, p. 12.
9. R.I. Geran, N.H. Greenberg, M.M. Macdonald, A.M. Schumacher, and B.J. Abbott, *Cancer Chemother. Rep., Part 3*, **3**(2), 1 (1972).
10. P.J. Ferguson, M.H. Fisher, J. Stephenson, D.H. Li, B.S. Zhou, and Y.C. Cheng, *Cancer Res.*, **48**, 5956 (1988).
11. K.H. Lee, Y.M. Lin, T.S. Wu, D.C. Zhang, T. Yamagishi, T. Hayashi, I.H. Hall, J.J. Chang, R.Y. Wu, and T.H. Yang, *Planta Med.*, **54**, 308 (1988).